

Stereoselective Synthesis of Spirocyclic Oxindoles via Prins Cyclizations

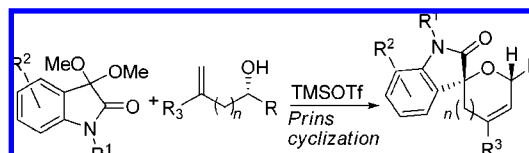
M. Paola Castaldi, Dawn M. Troast, and John A. Porco, Jr.*

Department of Chemistry and Center for Chemical Methodology and Library Development (CMLD-BU), Boston University, 590 Commonwealth Avenue, Boston, Massachusetts 02215

porco@bu.edu

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ABSTRACT



The synthesis of spirocyclic oxindole pyran and oxepene frameworks using highly stereoselective Prins cyclizations of homoallylic and *bis*-homoallylic alcohols and isatin ketals is described.

Exo-methylene pyrans are present in a variety of biologically active natural products.¹ We recently employed the intramolecular silyl-modified Sakurai (ISMS) reaction² to construct the exomethylene pyran subunit of the macrocyclic core of (–)-zampanolide (**1**, Figure 1a).³ We envisioned use of this powerful methodology to access a variety of exomethylene tetrahydropyrans (**3** and **4**, Figure 1b) using diverse ketals and acetals (**5** and **6**, Figure 1b) for diversity-oriented synthesis⁴ and chemical library development (Figure 1b). As part of our studies, we also considered preparation of pyran

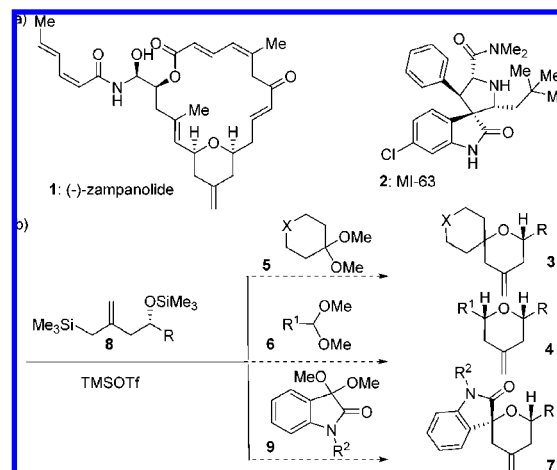


Figure 1. (a) Representative *exo*-methylene pyran spirooxindole molecules. (b) Initial synthesis plan.

spirooxindole hybrid molecules⁵ (**7**, Figure 1b) from allylsilanes **8** and isatin ketals **9** in order to merge fragments of two biologically interesting motifs. The spirooxindole core structure is represented in numerous pharmacological agents and alkaloids⁶ including the anticancer agent MI-63 (**2**, Figure 1a).⁷ In this communication, we report how our initial

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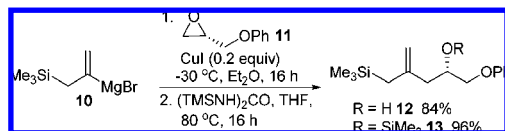
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synthesis plan evolved to identify stereoselective, Lewis acid-mediated Prins cyclizations to access both enantioenriched spirocyclic oxindole pyrans and oxepenes.

In initial studies, allyl silane **12** and derived silyl ether **13** (Scheme 1) were prepared employing Cu(I)-catalyzed⁸

Scheme 1. Preparation of Allylsilane **13**



ring-opening of chiral, nonracemic epoxide **11** with vinyl Grignard **10**.⁹ After optimization of ISMS reaction conditions, ketals **14** and acetal **15**¹⁰ were successfully employed in the synthesis of spirocyclic *exo*-methylene pyran **16** and 2,6-*syn*-disubstituted pyran **17**¹¹ (Table 1, entries 1 and 2).

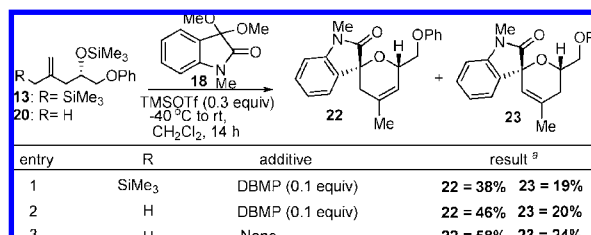
Table 1. Use of Allylsilane **13** in ISMS Reactions^a

entry	acetal/ketal	T [°C]	solvent	product	yield (%)
1		-78	Et ₂ O		80
2		-78	Et ₂ O		84 dr > 20:1
3		-78	Et ₂ O		<2 ^b
4		-78 to rt	Et ₂ O		63 dr = 1:1
5		-78 to rt	CH ₂ Cl ₂		36

^a Reaction conditions: allylsilane **13** (1.1 equiv), TMSOTf (0.3 equiv), 2,6-*t*-Bu-4-Me-pyridine (DBMP) (0.05 equiv), 4 Å MS, 14 h. ^a Isolated yields. ^b Not observed, *N*-methylisatin and desilylated allylsilane **20** recovered.

Subjection of *N*-methyl isatin dimethylketal **18**¹² (Table 1) and allyl silane **13** to optimized ISMS reaction conditions (Table 1, entry 3) did not afford the desired spiroannulated product **19** and led only to recovery of silyl ether¹³ **20** (Scheme 2) and *N*-methyl isatin. Warming of the reaction to room temperature afforded product **21**, presumably from direct allylation of the derived isatin oxonium ion with allylsilane **13** (Table 1, entry 4). Attempts to convert compound **21** into the desired spiroannulated product under acidic conditions afforded a complex mixture of products. When CH₂Cl₂ was used as solvent, spirooxindole **22** bearing

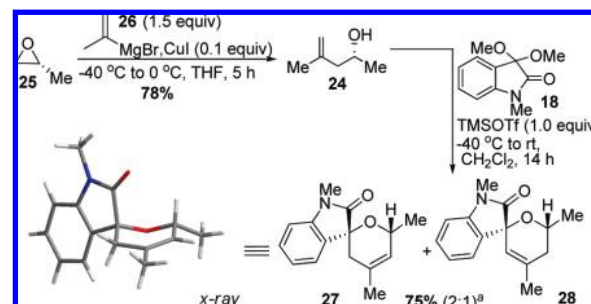
Scheme 2. Spirooxindole Pyrans via Prins Cyclizations



^a Isolated yield.

an endocyclic alkene¹⁴ was isolated in moderate yield. A control experiment employing ketal **14** in the ISMS reaction with CH₂Cl₂ as a solvent afforded product **16** and its endocyclic olefin isomer (1:2 ratio).

Scheme 3. Diastereoselective Synthesis of Spirooxindoles



^a Ratio of **27**:**28** determined by ¹H NMR analysis of a crude sample.

Based on these observations and given that spiroannulation did not occur at low temperature (−78 °C), we performed the reaction at higher temperatures which afforded spirooxindoles **22** and **23** in good overall yield (Scheme 2, entry 1). The apparent isomerization of the double bond (endocyclic vs exocyclic olefin) suggested the possibility of a mechanistic pathway which was different than the expected ISMS reaction. Intramolecular Prins-type cyclization¹⁵ of homoallylic silyl ether **20** derived from desilylation of **13** could account for such an outcome. To support this hypothesis, homoallylic silyl ether **20** was synthesized and employed in the reaction to afford products **22** and **23** in good overall yield (Scheme 2, entries 2 and 3). An additional spirooxin-

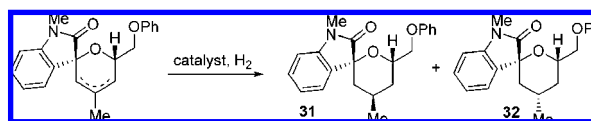
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Table 2. Amide-Directed Hydrogenation



spirooxindole	catalyst	H ₂	solvent	conversion ^a (%)	ratio ^a
22	Pd/C	50 psi	MeOH	96	2:1
23	Pd/C	50 psi	MeOH	99	7:1
22	Wilkinson's catalyst RhCl(PPh ₃) ₃	50 psi	EtOH/benzene		
22 + 23	Crabtree's catalyst [Ir(cod)py(PCy ₃)PF ₆]	1 atm	CH ₂ Cl ₂	99	only 32

^a Conversion and ratios of **31:32** determined by ¹H NMR analysis of crude samples.

dole pyran synthesis sequence is shown in Scheme 3. Preparation of homoallylic alcohol **24**¹⁶ by epoxide ring-opening, followed by treatment with isatin ketal **18** and TMSOTf, afforded spirooxindoles **27** and **28** (Scheme 3). The relative stereochemistry and alkene position of major stereoisomer **27** were confirmed by X-ray crystallographic analysis.¹¹

In order to explain the stereochemical outcome of the Prins cyclizations, we propose a chair transition state (Figure 2)

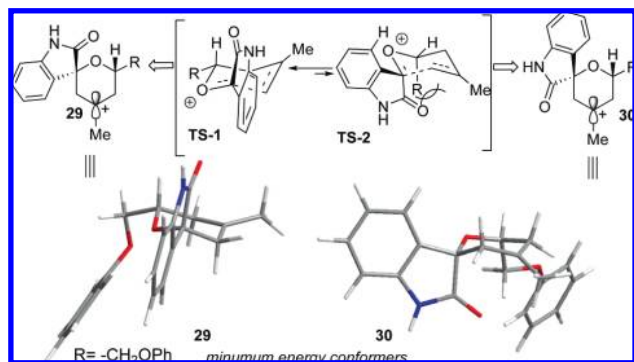


Figure 2. Proposed transition states.

in which the larger aryl substituent of the oxindole moiety adopts a *pseudo-equatorial* orientation¹⁷ (TS-1) leading to the observed diastereoisomer (cf. **22** and **23**, Scheme 2). An alternative chair (TS-2) leading to the disfavored diastereoisomer has significant steric interactions between the isatin carbonyl oxygen and the R substituent on the chiral center. Examination of molecular models of the proposed intermediate tertiary carbocations **29** and **30** obtained using Spartan conformational searches (AM1) followed by DFT minimization (performed using a 6-31G* basis set; $\Delta E = 8.25$ kcal/

mol)¹¹ shows destabilizing 1,3-diaxial interactions in carbocation **30** which is derived from Prins cyclization through TS-2.

In order to confirm the relative stereochemistry at the spiro center of minor regioisomer **23** generated during the Prins cyclization, we subjected both regioisomers **22** and **23** to metal-catalyzed hydrogenation. Interestingly, using catalytic amounts of Pd/C, a mixture of chromatographically separable diastereoisomers **31** and **32** were observed by ¹H NMR analysis of crude samples, indicating that regioisomers **22** and **23** had the same relative stereochemistry at the spiro center (Table 2).

In light of the poor diastereoselectivity observed using standard hydrogenation conditions, we next evaluated the possibility of amide-directed hydrogenation.¹⁸ While use of Wilkinson's catalyst did not generate the desired hydrogenated product, use of Crabtree's catalyst¹⁹ led to the production of **32** in excellent diastereoselectivity (dr >30:1) indicating complete substrate control in the amide-directed hydrogenation (Table 2).

In order to broaden the scope of the methodology to access spirocyclic oxindoles, we prepared a series of homoallylic alcohols (**24**, **33–36**) and isatin ketals (**18**, **37**,²⁰ **38**) for examination in the Prins cyclization (Table 3). Cyclizations were found to be successful with isatin ketals bearing NH functionality to afford spirooxindole products **39–43**. Introduction of a bulky bromine substituent on the 4-position of the isatin ketal (Table 3, entries 2, 4 and 5) resulted in improved diastereoselectivity and noticeably influenced the product olefin regiochemistry (cf. entries 3 and 4), which may be explained by highly regioselective elimination of a carbocation intermediate distal from the bromo-oxindole moiety (cf. **29**, Figure 2).

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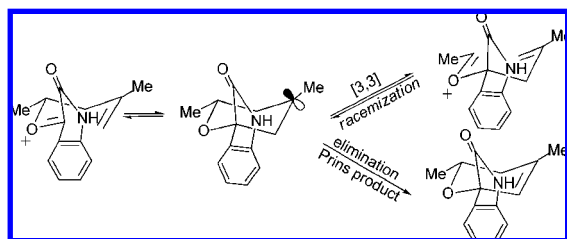
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Table 3. Prins-Type Spiro-annulation

entry	alcohol	isatin	product	yields (%) ^{a, c}
1				78 ^b dr = 13:1 rr = 2.6:1 ee > 99%
2				70 ^c dr > 20:1 rr = 13:1 ee > 99%
3				85 ^{b, d} dr > 10:1 rr = 4.3:1 ee > 99%
4				88 ^c dr > 20:1 rr = 10:1 ee > 99%
5				78 ^b dr = 20:1 rr = 10:1 ee > 99%

^a Isolated yields ^b Reaction conditions: TMSOTf (1.0 equiv), -40 °C to rt, 14 h, CH₂Cl₂. ^c TMSOTf (1.0 equiv), -40 to 0 °C, 3 h, CH₂Cl₂. ^d Isolated as an inseparable mixture of regioisomers. Hydrogenation was necessary to facilitate products separation. ^e rr = regioisomeric ratio; major isomer shown.

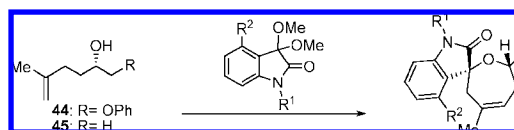
Considering the well-documented racemization observed during Prins cyclization due to competitive oxonia-Cope rearrangement (Figure 3),²¹ we also measured the enantio-


Figure 3. Possible racemization of Prins cyclization products via 2-oxonia-Cope rearrangement.

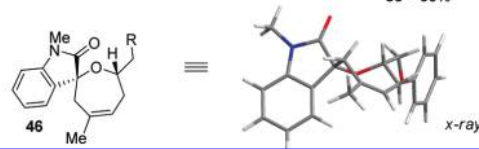
meric excess of the spirocyclic products. In all cases, we did not observed erosion in enantiopurity (Tables 3 and 4). These findings are consistent with the observations that stabilization of the intermediate tetrahydropyranyl cation raises the transition states energy for ring-opening and effectively eliminates the oxonia-Cope rearrangement.^{21a,b}

Finally, we extended the methodology to intramolecular Prins cyclization of *bis*-homoallylic alcohols (Table 4, **44**, **45**²²) and isatin ketals (Table 4, **18**, **37**, **38**) to generate spirocyclic oxindole oxepenes^{14b,23} **46–50** in high diastereo- and regioselectivity (Table 4). The relative stereochemistry of spirooxindole **46** was confirmed *via* X-ray crystallographic analysis.¹¹

In conclusion, enantiopure spirocyclic oxindole pyrans and oxepenes have been efficiently synthesized by highly ste-

Table 4. Diastereoselective Synthesis of Spirooxindole Oxepenes


entry	R	R ¹	R ²	product	yields (%) ^{a, b}
1	OPh	H	Me	46	51 dr > 20:1 rr = 10:1 ee > 99%
2	OPh	H	H	47	41 dr > 20:1 rr = 10:1 ee > 99%
3	OPh	Br	H	48	43 dr > 20:1 rr = 13:1 ee > 99%
4	H	H	H	49	56 dr > 20:1 rr = 16:1 ee > 99%
5	H	Br	H	50	57 dr > 20:1 rr = 20:1 ee > 99%



^a Isolated yields. ^b rr = regioisomeric ratio; major isomer shown. Reaction conditions: TMSOTf (1.0 equiv), -40 °C to rt, 3 h, CH₂Cl₂.

reoselective Prins-type cyclizations of both homoallylic and *bis*-homoallylic alcohols and isatin ketals. The protocol is highly complementary to related annulations involving chiral organosilanes.²⁴ Further studies involving related annulations and library synthesis applications are in progress and will be reported in due course.

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Supporting Information Available: Detailed experimental procedures and spectral data for all compounds. X-ray crystal structure coordinates and files in CIF format. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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